An electrolynamic integration ... 0/119/62/001/003/ 03/000 D201/0303

mass, m_2 - the mass of the coll) and a finite value of ., the frequency responde in this band becomes a structure of the abscisca as it is in a displacement pick-up. For , which higher irequencies the frequency response is distorte, contains the effect of m_2 . The following optimum values of prometers are recommended: $\alpha = 0.5$, p = 0.001; $\gamma = 1.5$ (γ is lambling in the presence of a stiff link k_2). The above values correspond to a flat frequency characteristic of the pick-up shifted towards low frequency as much as possible. Good mamping of tree oscillations of the pick-up is obtained from the current flowing in the neasuring circuit. There are) figures and 2 Soviet-bloc references.

Card 2/2

S/194/61/000/010/021/082 D222/D301

AUTHOR:

Pikulev, N.A.

TITLE:

Some problems in the electrical modelling of the passage through resonance of vibration-isolated

machines or foundations

PERIODICAL:

Referativnyy zhurnal. Avtomatika i radioelektronika, no. 10, 1961, 17, abstract 10 Blo9 (Tr. Tsentr. n.-i. in-ta stroit. konstruktsii. Akad. str-va i arkhitekt

SSSR, 1961, no. 1, 104-115)

The specific conditions of constructing electrical models for mechanical oscillatory systems under the influence of forces of changing frequency and amplitude are considered. Some analogue devices are described which can be used in building analogues for the study of phenomena taking place when a vibrator-iso-lated machine or foundation goes through resonance. 8 figures. 10 references. [Abstracter's note: Complete translation]

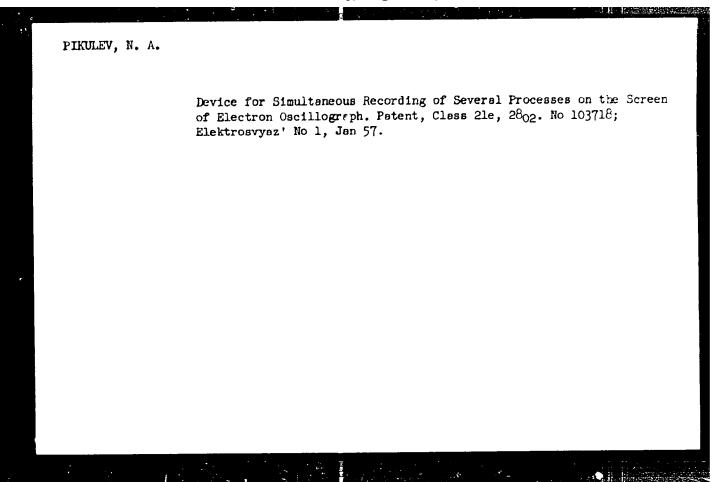
Card 1/1

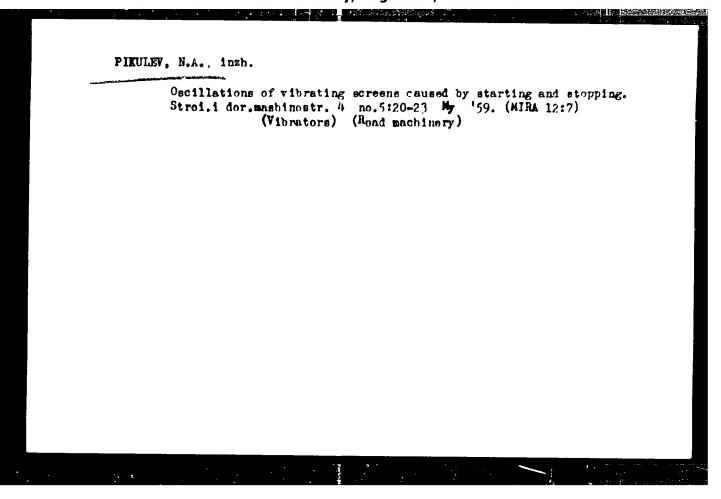
PIKULEV, N. A., Cand Tech Sci -- (diss) "Work of vibration insulation under conditions of starting and stopping of machines. Methods of devinder conditions of starting and stopping of machines. Methods of decreasing resonance vibrations." Moscow-Sverdlovsk, 1960. 20 pp; (Acadcresing resonance vibrations." Moscow-Sverdlovsk, 1960. 20 pp; (Acadcresing resonance vibrations under USSR, Central Scientific Research emy of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Structures, Scientific Research Inst of Industrial Inst of Construction Inst of Construction Inst of Industrial Inst of Construction Inst of Constructio

Role of elastic pieces inserted between dampers and the vibrating body. Stroi.mekh. i rasch.soor. 1 no.2:45-48 '59. (MIRA 12:4)

(Machinery--Vibration) (Damping (Mechanics))

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240





PIKULEV, N.A., inzh. Reducing vibrations of huildings. Masl.-zhir.prom. 25 no.10: 45-46 '59. 1. Nauchno-issledovatel'skiy institut promedaniy i soorusheniy Akademii stroitel'stva i arkhitektury SSSR. (Buildings--Vibration) (Compressors)

J. H. Bray Mannager Land

PIKULEVA V.A.

Pood of Hemibarbus maculatus Bleek. and Hemibarbus labeo Pall. in the Amur Basin. Mat. k pozn. fauny i flory SSER. Otd. sool. no.32:419-434 *52. (MIRA 11:4)

1. Laboratoriya ikhtiologii Moskovskogo gosudarstvennogo universiteta.

(Amer Valley-Carp) (Fishes-Food)

PIKULEVA, Yu.V., kand.med.nauk

Reactive changes in the organism in osseous tuberculosis during morbid and healing processes [with summary in French]. Probl.tub. (MIRA 12:2) 37 no.1:27-31 159.

1. Iz Instituta tuberkuleza AMN SSSR (dir. Z.A. Lebedeva; zav. klinikoy kostnosustavnogo tuberkuleza imeni T.P. Krasnobayeva - prof. Z.Yu. Rol'ye).

(TUBERCULOSIS, OSTEOARTICULAR, immunol.

phagocytosis in morbid & healing processes (Rus))

(PHAGOCYTOSIS, in var. dis.

tuberc. osteoarticular, during morbid & healing
processes (Rus))

HIROL'SKIY, G.V.; PIKULEVA, V.A.

Adaptive significance of the amplitude of variability of specific characters and properties of the organism [with summary in English]. Zool. shur. 37 no.7:972-988 Jl '58. (MIRA 11:8)

1. Laboratoriya ikhtiologii Moskovskogo gosudarstvennogo universiteta.

(Adaptation (Biology)) (Fishes--Food)

Butrition of Pseudobagrus fulvidraco Rich, and Idocassis brashnikovi Berg, in the Amur Resin, Mat. k pozn fauny i flory SSSR. Otd. zool. no.32:435-448 *52. (MIRA 11:4) 1. Iaboratoriya ikhtiologii Moskovekogo gosudarstvennogo universiteta. (Amur Valley—Catfishes) (Fishes—Food)

HIKOL'SKIY, G.V.; GROMCHEVSKAYA, N.A.; MOROZOVA, G.I.; PIKULEVA, V.A.

Pishes of the upper Pechora basin. Mat.k pezn.favay i flory SSSR.Otd., zeol.no.6:5-202 '47. (NURA 9:9)

(Pechora River--Pishes)

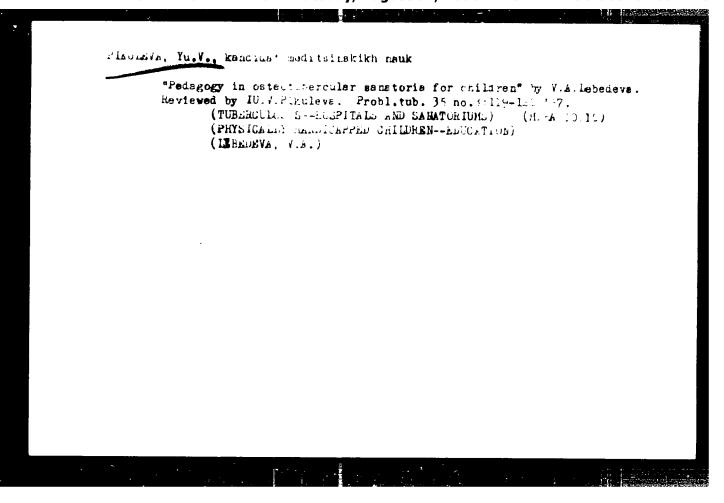
一次是任命和关系的统治的关系是

PIKULEVA, Yu.V.

Dystrophic changes in tissues in tuberculous coxitis in children [with summary in French]. Probletube 36 no.3:47-53 '58 (MIRA 11:5)

Iz kliniki kostno-sustavnogo tuverkuleza imeni T.P. Krasnobayeva (zav. - prof. Z.Yu. Rol'ye) Instituta tuberkuleza AMN SSSR (Dir. Z.A. Lebedeva)
 (TUBERCULOSIS, OSTEOARTICULAR, in inf. & child dystrophic changes in tissues in coxitis (Rus))

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240



THE RESIDENCE OF THE PROPERTY OF THE PROPERTY

PIKULEVA, Yu.V., kand. med. nauk

Experience in the use of steroid hormones in children with various forms of osteoarticular tuberculosis. Probl. tub. no.4:61-66 164. (MIRA 18:11)

1. Otdeleniye kostno-sustavnogo tuberkuleza imeni T.P. Krasnobayeva (zav. - prof. Z.Yu. Rol'ye) TSentral'nogo instituta tuberkuleza (direktor - deystvitel'nyy chlen AMN SSSR prof. N.A. Shmelev) Ministerstva zd. avookhraneniya SSSR, Moskva.

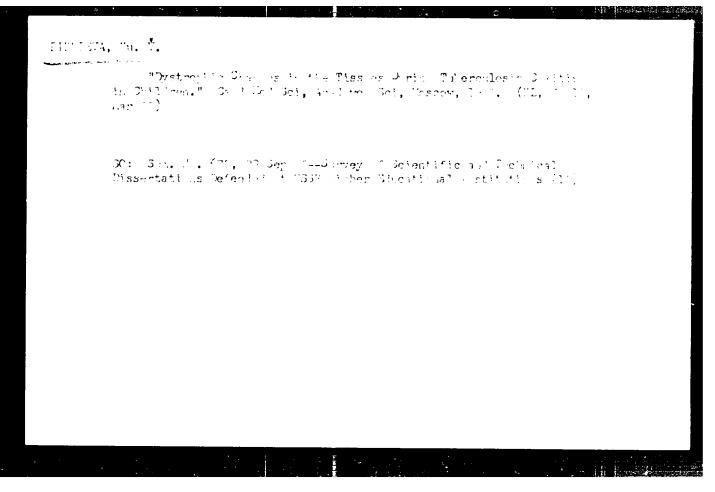
PIKULEVA, Yu.V., kand.med.nauk

Case of a typical tubercular process in the hip joint. Probl. tub. no.8.107-109'62. (MIMA 16:9)

l. Iz kliniki kostno-sustavnogo tuberkuleza imeni T.P. Krasnobayeva (zav. klinikoy - prof. Z.Yu.Rol'ye).

(HIP JOINT-TUBERCULOSIS)

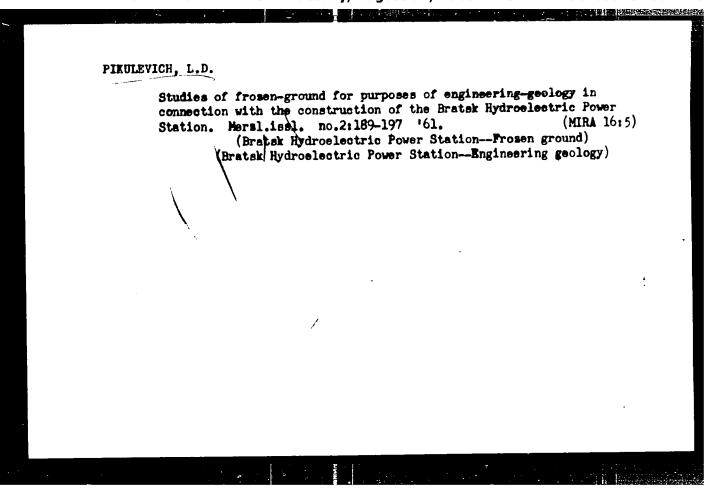
"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240



"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240

Soil heaving in the region of the Bratak Hydroelecter Domes Station, Merzl. issl. no.3.131-157 to3.

Stages in freezing processes and the variation of moisture in seasonally frozen ground. Ibid. 152-168 (MEGA 17.1)



PIKULIK, Alojz, inz.

Developing the rescue service in the mines of Slovakia. Uhli 5 no.1:26 Ja '63. Uhli 5 no.1:26 Ja '63.

1. Hlavni banska sachranna stanica, Handlova.

AP4043007 ACCESSION NR:

8/0051/64/017/002/0209/0212

Gladchenko, L. F.; Pikulik, L. G.; Belosarevich, M. L.

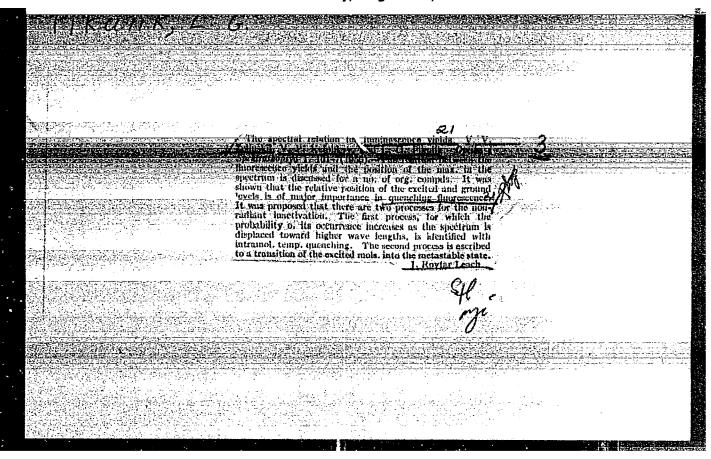
TITLE: Study of electron spectra of a series of aromatic molecules in solutions

SOURCE: Optika i spektroskopiya, v. 17, no. 2, 1964, 209-212

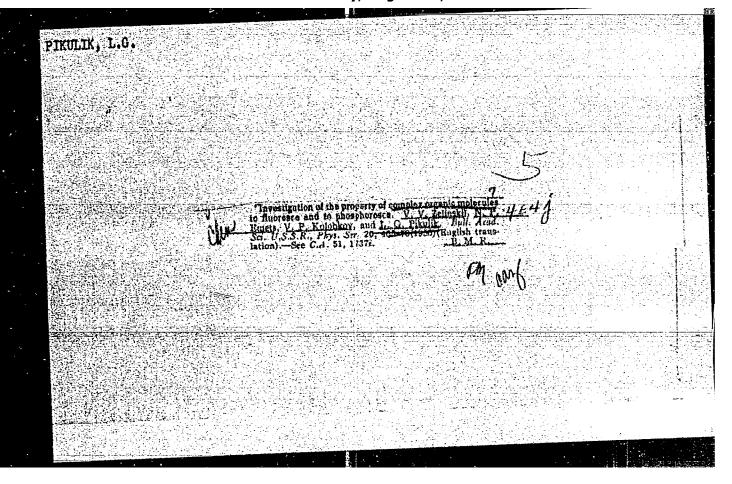
TOPIC TAGS: aromatic compound spectrum, electron transition, luminescence, fluorescence, phthalimode, benzene

ABSTRACT: In order to obtain a more reliable determination of the characteristic electron-vibration frequencies in each spectral band, and to trace their variation from compound to compound, the absorption spectra of benzene, toluol, benzoic acid, phthalic acid, phthalic anhydride, and phthalimide were measured at low temperatures using solvents that either vitrify or crystallise upon cooling (ethyl alcohol and dioxane, respectively). Analysis of the data shows that each of the spectra of all the substances contain funda-

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240



"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240



PIKULIKI L. G. Chemical Bonds. USSR/Physical Chemistry - Molecules. Chemical Bonds.

Abs Jour: Ref Zhur-Khimiya, No 5, 1957, 14384

: V. V. Zelinskiy, N. P. Emets, V. P. Kolobkov, L. G. Author

Pikulik

Investigation of the capacity of complex organic mole-Inst Title

cules to fluoresce and phosphoresce

Izv. AN SSSR, ser. fiz, 1956, 20, No 5, 507-513 Orig Pub:

Abstract: An investigation was made of the dependence of the probability of non-radiating transitions of excited molecules (from the unstable level to the metastable level r, from the unstable to the basic without q radiation, from the metastable to the basic with T radiation, and from metastable to the basic without radiation q2) on the temperature, solvent and molecule structure. Probability of r is apparently only weakly dependent on temperature. Probability q changes little with temperature for some organic compounds while for

Card 1/2

PIKULIK, L. G., Card Phys-Math Sci -- (diss) "Effect of the Solvent upon the Quantum butput of Fluorescence and Electronic Spectra of Flatimids ." Minsk, 1957. 10 pp (Acad Sci Belorussian SSR, Department of Physicomathematical and Technical Sci), 100 copies (KL, 48-57, 104)

A Jest March and photological

- 6 -

Fire Like L. G

AUTHORS: Zelinskiy, V.V., Kolobkov, V.P. and Pikulik, L.G. 51-3-23/24
TITLE: Dependence of the fluorescence and absorption spectra on

TITLE: Dependence of the little state and address the solvent for certain phthalimide derivatives.

the solvent for certain phonomia is pogloshcheniya ot (Zavisimost' spektrov fluorestsentsii i pogloshcheniya ot

rastvoritelya u nekotorykh proizvodnykh ftalimida).

PERIODICAL: "Optika i Spektroskopiya" (Optics and Spectroscopy), 1957, Vol.2, No.3, pp.402-405 (U.S.S.R.)

ABSTRACT: The effect of the solvent on the fluorescence and absorption spectra of phthalimide derivates consists of a displacement of the spectral band without alteration of its shape.

This paper presents experimental results at room temperature for the 4-derivates (8 compounds such as 4-aminophthalimide), for the 4-derivates (10 compounds such as 3-acetylamino-N-methyl-3-derivatives (12 compounds such as 3-hacetylamine-N-methylphthalimide) of phthalimide.

3,6-diacetylamine-N-methylphthalimide) of phthalimide.

The maximum frequency of fluorescence or absorption is plotted against the solvent (23 solvents were used, e.g. water, glycerin, methanol, pyridine, acetone, CCl₄, etc). The glycerin, methanol, pyridine, acetone, CCl₄, etc). The solvents are given places on the abcissa in such a way that distances are proportional to the spectral displacement distances are proportional to the spectral displacement produced by that particular solvent on a standard substance which is 4-amino-N-methylphthalimide. For the fluorescence

Dependence of the fluorescence and absorption spectra on the solvent for certain phthalimide derivatives. (Cont.) spectral displacement sets of straight lines are obtained, for the absorption spectra the dependence is more complex. The effect of the solvent could not be correlated with its dielectric constant or its dipole moment and further work, e.g. on variation of the spectral displacement with temperature, is suggested.

There are 3 figures and 8 references, 7 of which are Slavic.

SUBMITTED: October 22, 1956.

AVAILABLE:

Card 2/2

PIKULIK, L.G

AUTHORS: Zelinskiy, V.V., Kolobkov, V.P. and Pikulik, L.G. 51-3-24/24

An attempt at construction of a unidimensional model of potential curves for certain derivatives of phthalimide. TITLE:

(Popytka postroyeniya odnomernoy modeli potentsial'nykh

krivykh dlya nekotorykh proizvodnykh ftalimida).

PERIODICAL: "Optika i Spektroskopiya" (Optics and Spectroscopy), 1957, Vol.2, No.3, pp. 405-408 (U.S.B.R.)

ABSTRACT: Starting from the absorption and fluorescence spectra, the authors construct unidimensional potentials for complex molecules. It is assumed that electron transitions do not violate the Franck-Condon principle. The transitions from various points of a potential curve are taken to be equally probable. The distribution of molecules in various vibration levels at a given temperature is taken to be the same for different molecules and for both the ground and excited states. The method of construction of these potentials is described and potential curves are shown for the ground and excited states of 4-amino-N-methylphthalimide solutions in methyl alcohol and pyridine, and of 3,6-diacetylamino-N-methylphthalimide solutions in ethyl alcohol and benzene. When the absorption and fluorescence spectra are mirror-symmetrical Card 1/2 the potentials for the ground and excited states are identical,

PIKULIK, L.C.

51-5-24/26

AUTHOR:

TITIE:

On the Effect of a Solvent on the Luminescence Yield. Pikulik, L.G. (0 vliyanii rastvoritelya na vykhod fluorestsentsii)

Optika i Spektroskopiya, 1957, Vol.2, No. 5,

PERIODICAL: pp. 676 - 678 (USSR)

To study the effect of a solvent on the quantum yield of fluorescence, the yield of vapour and solution should be measured at the same temperature. The author uses glycerin (which ABSTRACT: boils at 280°C) as a solvent. The experiment shows that on variation of temperature of the solutions of phthalimides in glycerin from 20 - 270°C, both a displacement of fluorescent spectrum and change in the quantum yield occur. For 3-aminospectrum and change in the quantum yield occur. For 3-aminophthalimide (Fig.1), the displacement in the short-wavelength direction is 900 cm⁻¹, for 3-monomethylaminophthalimide (Fig.2), the displacement is 800 cm⁻¹. At the same time, the band widens; in Fig.1 from 3600 to 4250 cm⁻¹; in Fig.2, from 4200 to 4700 cm⁻¹. The dependence of the yield on temperature is shown in Fig. 3, where I refers to 3-aminophthalimide and II refers to 3-monomethylaminophthalimide. In both cases. A II refers to 3-monomethylaminophthalimide. In both cases, a fall in the yield occurs on increase of temperature from 60 to 270°C. At 220°C, the yield is 34% for I and 4.8% for II and at the same temperature when vapours are excited at a purely

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CIA-RDP86-00513R0012408 APPROVED FOR RELEASE: Tuesday, August 01, 2000

51-5-24/26

On the Effect of a Solvent on the Luminescence Yield.

electron frequency, the yield is 1.2% for I and 0.2% for II. Thus, the yield of the solution of I is 28 times larger than the yield of its vapour; for II the solution yield is 24 times larger than the vapour yield. The results obtained both at 220 °C and at 20 °C show that the solvent increases the fluorescence yield. This effect is due to displacement of the emission band by the solvent which materially affects quantum yield. [Refs. 8, 9]. The author thanks V.V. Zelinskiy for suggesting the investi-

gation and A.N. Sevchenko and B.I. Stepanov for advice.

IATION: Institute of Physics and Mathematics Ac.Sc. Belorussian SSR. (Institut Fiziki i Matematiki AN BESH)

SUBMITTED:

January 15, 1957.

AVAILABLE:

Library of Congress.

Card 2/2

SOV/48-22-11-26/33

24(7) AUTHORS: Pikulik, L. G., Solomakho, M. A.

TITLE:

Temperature Dependences of the Electron Spectra in Some Phthalimides (Temperaturnyye zavisimosti elektronnyki.

spektrov u nekotorykh ftalimidov)

PERIODICAL:

Izvestiya Akademii nauk SSSR. Seriya fizicheskaya, 1958,

Vol 22. Nr 11, pp 1391-1394 (USSR)

ABSTRACT:

Considerable displacements in the emission spectra of Ththalimide solutions have been observed, if the temperature of the medium was varied from +20° to +250° (Refs 2, 3) ture of the medium was varied from +20° to -196° (Ref 4). In order to determine the and from +20° to -196° (Ref 4). causes for such considerable displacements the authors investigated the absorption- and emission spectra of some Inthalimides in high-boiling solvents. The temperature influence of the medium upon the position of the band is dependent upon their initial position as determined by the respective solvent. If the temperature of the medium is raised, the phthalimide spectra may suffer a considerable displacement, if they were, with respect to their position in the vapor, much dis laced previous to heating. The ef-

Card 1/3

sov/48-22-11-26/33

Temperature Dependences of the Electron Spectra in Some Phthalimides

fectiveness of the temperature-dependent band displacement is, for a certain substance, dependent upon the position of the band at room temperature. This influence may primarily be ascribed to modifications of the influence of the solvent upon the molecule. The regularities in question only apply to a certain limited temperature range. The temperature displacement of the phthalimide spectra can be described qualitatively in the range of from +20° to -196° by a onedimensional rotential curve model. The spacing between the minima of the potential curve determines the frequency of the electron transition. A reduction of the differences between the equilibrium radii of the upper and the lower electron state correspond to a mutual displacement of the absorption- and emission lines, if the temperature of the medium is varied. The absorption maximum evidently shifts toward lower and the luminescence maximum toward higher frequencies. According to the scheme the frequency of the electron transition should remain unchanged in such a case. In order to elucidate this question further investigations are required, which should cover the absorption spectra at low temperatures and in different media. The authors express

Card 2/3

SOV/48-22-11-26,'33

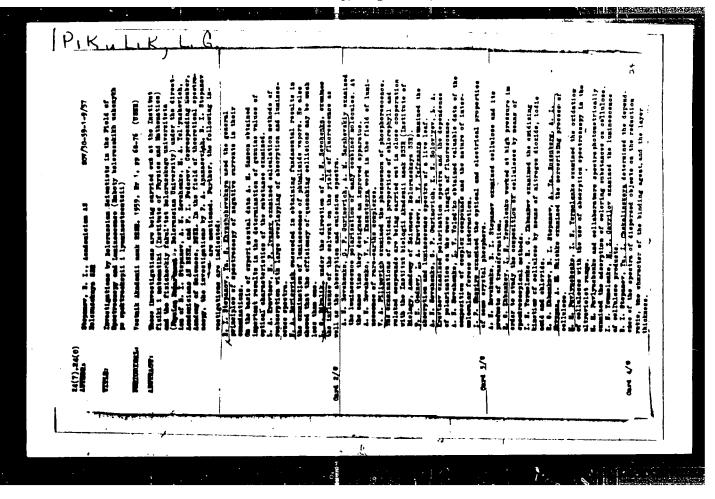
Tem; erature Dependences of the Electron Spectra in Some Phthalimides

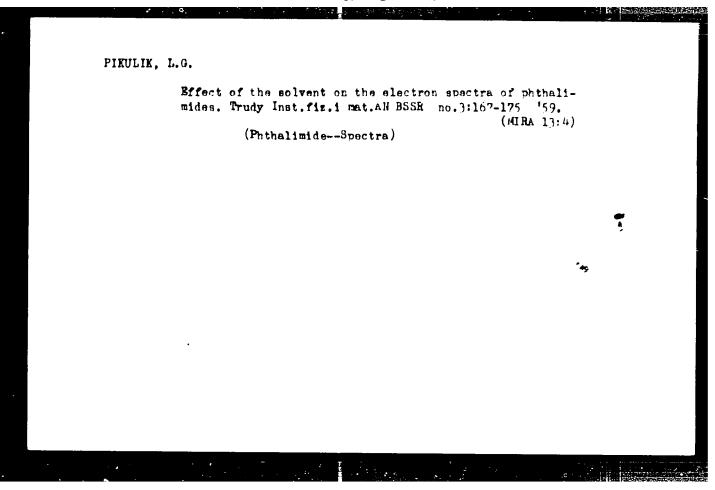
their gratitude to V. V. Zelinskiy, A. N. Sevchenko and B. I. Stepanov for valuable suggestions. There are 5 figures and 8 references, which are Soviet.

ASSOCIATION: Institut fiziki i matematiki Akademii nauk BSSR

(Institute of Physics and Mathematics, AS Belorussian SSR)

Card 3/3





AUTHORS:

Pikulik, L. G., Solomakha, M. A.

Bo19/B054 82322

Pikulik, L. G., Solomakha, M. A.

Investigation of the Temperature Influence on the Luminescence- and Absorption Spectra of Complex

Molecules in Solutions

Molecules in Solutions

PERIODICAL: Inzhenerno-fizicheskiy zhurnal, 1960. Vol 3, No. 6, p. 53-60

TEXT: To clarify the causes of the shift of solution and emission the absorption- and emission the authors investigated the absorption materials at the

TEXT: To clarify the causes of the shift of solution and emission the meratures the authors investigated the absorption—and solutions spectra of phthalimides and of some typical coloring materials at the spectra of phthalimides and solutions. They studied liquid and solid solution—and temperature of liquid nitrogen. They studied liquid and solid solution—and temperature of some phthalimides as a function of temperature. The experimental arrangement is briefly described, the absorption—and the experimental arrangement is briefly described, the absorption—and the experimental arrangement is briefly described, the absorption—and temperature in fig. 3 show that the maxima of the absorption of the bands in dependence on fluorescence bands approach one another with decreasing in dependence of the bands in dependence of the maxima of the bands in dependence of the bands in dep

Card 1/3

APPROVED FOR RELEASE: Tuesday, August 01, 2000

CIA-RDP86-00513R0012408

Investigation of the Temperature Influence on the 5/170/60/003/008/006/014B019/B054 82322 Luminescence- and Absorption Spectra of Complex Molecules in Solutions

phthalimides have the same character of band shifting. In the solutions mentioned at first (Fig. 3, Table 1) the optical transparency is maintained at the change in state caused by the cooling whereas in the solutions with different character of shifting the strong fissuring exerts an influence on the change of state Further investigations have shown that the shift of the fluorescence- and absorption spectra occurs in the direction of the electron transition frequency. This effect may be observed both with solid solutions and with liquid ones which change their state with decreasing temperature. Theoretical investigations show that the spectral shift toward the electron transition frequency is due to the change in probability of vibrational transitions of electrons. This is explained in connection with the increasing effect of the medium on the fluorescent molecule. B. I. Stepanov (Ref. 1), L. P. Kazachenko (Ref. 7), and R. V. Lel'chuk are mentioned. The authors thank A. N. Sevchenko and B I Stepanov, Academicians of the AS BSSR, for valuable advice, and V. V. Zelinskiy, Candidate of Chemical Sciences, for the supply of the phthalimides. There are 4 figures, 1 table, and 15 Soviet references.

Card 2/3

Investigation of the Temperature Influence on the S/170/60/003/00-80/051/3R0012408

Luminescence- and Absorption Spectra dust 01, 2000 B019/165/RDP86-0051/3R0012408

APPROVED FOR RELEASE1uTuesday, August 01, 2000 B019/165/RDP86-0051/3R0012408 82322

ASSOCIATION:

Institut Fiziki AN BSSR, g Minsk (Institute of Physics

SUBMITTED.

July 7, 1959

PIKULIK, L.G.; SOLOMAKHA, M.A.

Effect of temperature on the luminescence and absorption spectra of complex molecules in solutions. Inzh.-fiz.zhur. no.8: 53-60 kg '60. (MIRA 13:9)

1. Institut fiziki AN BSSR, g. Minsk.
(Luminescence) (Spectrum, Molecular)

5.3100

69839

S/051/60/008/03/010/038

AUTHORS: Pikulik, L.G., and Solomakho, M.A.

of Complex Molecules

PERIODICAL: Optika i spektroskopiya, 1960, Vol 8, Nr 3, pp 338-341 (USSR)

ABSTRACT: The authors investigated the absorption and fluorescence spectra of a number of phthalimides and of the following dyes: acridine yellow, suramine, chrysoidine, coryphosphine, fluorescein and aminochloromaleinimide. These substances were investigated both at 20 °C and at the liquid-nitrogen temperature. Ethyl, propyl and isobutyl alcohols were used as solvents. Solid solutions based on gelatine, sugar and starch, and films with polymethylmethacrylate and nitrocellulose bases were also studied. The fluorescence spectra were recorded with a photoelectric spectrometer based on a glass monochromator UM-2; the absorption spectra were recorded with a Card spectrometer SF-4. The absorption and fluorescence 1/2 spectra of 3-monomethylaminophthalimide and 4-aminophthalimide at 20 and -196 °C are shown ir.

69839

s/051/60/008/03/010/038

On the Effect of Temperature on the Electronic Spectra of Complex

Figs 2 and 3 respectively. It was found that in these two and in other compounds a lowering of temperature produced a displacement of the absorption and fluorescence maxima towards the frequency of a purely electronic transition. These displacements indicate a change in the probabilities of electron-vibrational transitions. change is due to the solvent which, on lowering of temperature, interacts more strongly with the solute molecule. In absorption the relative number of transitions to the lower vibrational levels of the upper electronic state increases on lowering of temperature, and the absorption maximum is displaced towards lower frequencies; in luminescence the number of transitions to the lower vibrational levels of the ground electronic state increases with lowering of temperature and the fluorescence maximum is displaced towards higher frequencies (Fig 1). Acknowledgements are made to A.N. Sevchenko and to B.I. Stepanov for their advice, and to V.V. Zelinskiy for

Card 2/2

the supply of the phthalimides.

There are 3 figures and 10 Soviet references. Submitted: ς, .

S/048/60/024/006/020/030/XX B013/B067

AUTHORS:

Pikulik, L. G. and Sevchenko, A. N.

TITLE:

Temperature Dependences of the Fluorescence Quantum Yieli

of Some Phthalimides in Various Solvents

PERIODICAL:

Izvestiya Akademii nauk SSSR. Seriya fizicheskaya. 1060,

Vol. 24, No. 6, pp. 729-733

TEXT: The authors studied the temperature dependences of the fluorescence yields of some phthalimides in high-boiling solvents. The student L. Vclodi-ko took part in the measurements. These solvents showed a considerable shift of the spectral lines when temperature was increased. The studies were made on a photoelectric apparatus in which an YM-2 (UM-2) meanor chromator was used. A spectrophotometer of the type ($\Phi-4$ (SF-4) was used to study the absorptiveness of the solutions. Fig. 2 shows fluorescence spectra of 4-dimethylamino-N-methylphthalimide in various media at different temperatures. In benzyl alcohol and cyclohexanol solutions, the quantum yield is increased by an increase of temperature, whereas in dimethylphthalate solutions it is reduced. Fig. 3a shows the temperature

Temperature Dependences of the Fluorescence S/048/60/024/006/020/030/XX Quantum Yield of Some Phthalimides in B013/B067 Various Solvents

dependence of the quantum yield of 4-amino-N-methylphthalimide. In benzy. alcohol, glycol, and cyclohexanol, the quantum yield is increased with rising temperature. In dimethyl phthalate, a thermal extinction of fluorescence takes place. This extinction of fluorescence was qualitative. ly observed in those solvents which have the peaks of their fluorescence lines in the region of $20,200 \div 24,000$ cm⁻¹. In the above-mentioned media, a similar change of the quantum yield was observed with 4-aminophthalimide. The temperature-dependent changes of the quantum yield observed may be explained by the relation $B_{quant} = f(v_{max}^{fl})$. According to this relation, a high value of the yield must correspond to the high value of $v_{\rm max}^{\rm fl}$ in the region of the ascending branch. A relationship between yield and spectrum was observed also in these cases where the spectrum was shifted not by the variable of the solvent but by the temperature change. With the maximum frequency of the fluorescence spectrum being shifted at 20°C toward higher frequencies, the yield is slightly increased. 3-amino-N-methylphthalimide and 4-amino-N-methylphthalimide have almost the same relations

Card 2/4

Temperature Dependences of the Fluorescence Quantum Yield of Some Phthalimides in Various Solvents

S/048/60/024/006/020/030/XX B013/B067

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"APPROVED FOR RELEASE: Tuesday, August 01, 2000

CIA-RDP86-00513R001240

Temperature Dependences of the Fluorescence Quantum Yield of Some Phthalimides in Various Solvents S/048/60/024/006/020/030/XX B013/B067

the yield and the spectrum of fluorescence. The authors thank <u>V. V. Zelinskiy</u> for valuable comments, and <u>A. M. Bonch-Bruyevich and G. A. Tishchenko</u> for making the fluorometric measurements possible. The present paper was read at the <u>Eighth Conference on Luminescence</u> (Molecular Luminescence and Luminescence Analysis) which took place in Minsk from October 19 to 24, 1959. There are 3 figures, 1 table, and 8 Soviet references.

ASSOCIATION: Institut fiziki Akademii nauk BSSR (Institute of Physics of the Academy of Sciences BSSR)

Card 4/4

L 9876-66 EWI (1)/ENI (m)/T LIP(c) RM

SOURCE COD: UR/0250/65/009/010/06/7/0650

AUTHOR: Gladchenko, L. F.; Kostko, M. Ya.; Fikulik, L. G.; Savchenko, A. K.

OKG: IPARB

DKG: Institute of Physics AN BSSN (Institut fisiki AN BSSN)

TITLE: Duration of the excited state of ultraviolet fluorescence of aromatic scide; 74,55

SOURCE: AN BSSN. Doklady, v. 9, no. 10, 1965, 640-650

TOFIC TAGS: aromatisation, smino acid, fluorescence

ABSTRACT: Mirect measurements of the duration of fluorescence of indole, tryptophan, glycyltryptophan, tyrostae, and physicalanine were carried out by using a supertrum. Distilled H20 and high purity EtOH were used as solvents. Known processing in the ultraviolet range of the formulas were used for calculation of the actual duration of fluorescence (7 act).

L 9876-66

ACC NR: AF5027349

Control measurements of previously known T were in good agreement (accuracy ~ 10%) with the values given in the literature. The optical characteristics derivations termined for the amino acids tested are given. Originat. has: I table.

SUB CODE: 07/ SUBM DATE: 22May65/

NR REF SOV: 002/ OTHER: 006

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2/2

L 33592-66 EWT(m)/EWP(j) RM

ACC NR: AR6016186 SOURCE CODE: UR/0058/65/000/011/D021/D021
APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R0012408
AUTHORS: Gladchenko, L. F.; Pikulik, L. G.; Belozarevich, N. L.

TITLE: Study of electron spectra of a series of aromatic molecules in solution

SOURCE: Ref. zh. Fizika, Abs. 11D151

REF SOURCE: Tr. Komis. po spektroskopii. AN SSSR, t. 3, vyp. 1, 1964, 319-326

TOPIC TAGS: electron spectrum, nonmetallic organic derivative, absorption spectrum, complex molecule

ABSTRACT: The authors investigated the electronic absorption and reflection spectra of a number of very simple derivatives of benzene and phthalimide in solutions at room that nower temperatures. The character of the long-wave absorption of phthalimide is analyzed on the basis of the change in the long-wave absorption band in a number of substances with different substitutent groups. The Stepanov universal relation is applied to the absorption and emission spectra of frozen solutions of complex molecules. [Franslation of abstract]

SUB CODE: 20

Cord 1/1 7

GLADCHENKO, L.F.; KOSTKO, M.Ya.; PIKULIN, L.G.; SEVCHENKO, A.N.

Duration of the excited state of ultraviolet fluoreaces e of aromatic amino acids. Dokl. AN BSSR 9 no.10:64/-650 0 '65.

(MTRA 18:12)

1. Institut fiziki AN BSSR. Submitted May 22, 1965.

ENT(1)/ENT(m)/ENP(j) UR/0020/65/162/001/0057/0059 L 1198-66 ACCESSION NR: AP5013440 11,55 AUTHOR: Sevchenko, A. N. (Academician AN BSSR); Pikulik, L. G.; Kostko, H. TITLE: Duration of fluorescence in complex molecules SOURCE: AN SSSR. Doklady, v. 162, no. 1, 1965, 57-59 TOPIC TAGS: fluorescence spectrum, luminescence, quantum yield ABSTRACT: The authors study the persistence of fluorescence in solutions as a spectral function of the frequency of the activating light including the anti-Stokes excitation region. Phthalimide derivatives and other organic compounds were used as study specimens. Water, alcohols and dioxane were used as solvents. A fluorometer was used with a measurement range of 2.10 1910 7 sec. The other equipment and the methods used in the experiment are described briefly. Special precautions were taken to reduce stray light which might fall onto the photomultiplier. The measurements were made at various concentrations depending on the intensity of the illumination. Since concentration quenching of fluorescence does not take place in phthalimides, a concentration close to saturation could be used where necessary. Card 1/2

L 4198-66

ACCESSION NR: AP5013440

It is found that the duration of luminescence is constant throughout the entire Stokes region of the excitation spectrum. Since the duration of the excited state is proportional to quantum yield for phthalimides and dyes under ordinary conditions, these data confirm Vavilov's law on a constant quantum yield in the Stokes excitation region. A constant fluorescence duration was also observed the anti-Stokes excitation region which indicates that quantum yield is also that in this region. A reduction in fluorescence duration was observed in the longer-wave anti-Stokes excitation region which should be studied further in view of the sharp reduction in absorption in this region and the consequent attentuation of luminescence which affects the reliability of measurements. In conclusion, the authors are deeply grateful to V. I. Shirokov, for making the control measurements and for assistance in adjusting the fluorometer, and also to V. V. Zelinskiy for furnishing the phthalimides. Orig. art. has: 2 figures, 1 table.

ASSOCIATION: Institut fiziki Akademii nauk BSSR (Institute of Physics, Academy of Sciences, BSSR)

SUBMITTED: 23Dec64

ENCL: 00

SUB CODE: OP, GC

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Card 2/2 01

OTHER: 000

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240

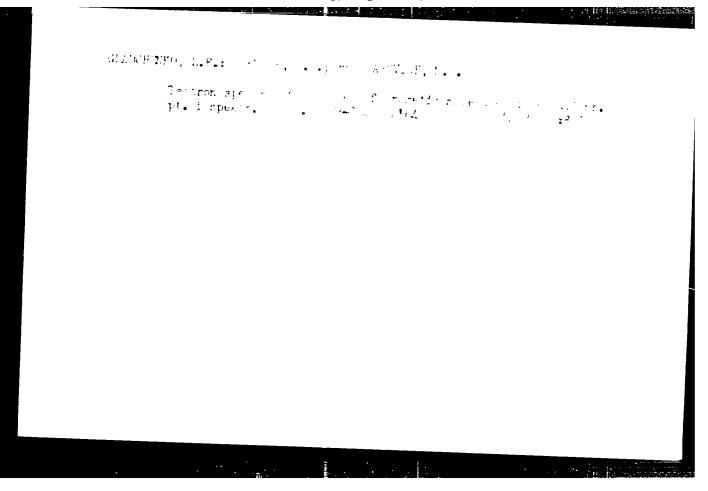
inner) i Leasile Hilbrig (1981).

PIKULIK, L.G.; GLADCHENKO, L.F.

Determining the constants of the sipple moments of a loss to t Dokl. AN BSCR 8 no.10:641-644 0 64. (Milli 18:3)

1. Irstitut fiziki AN BSSR.

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240



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THE RESERVE AND A STREET

KORIN, M.M.; MASHENKOV, V.A.; PIKULIK, L.G.

Ultraviolet absorption spectra of the cerebrospinal fluid in neurological patients. Dokl. AN BSSR 7 no.7:498-500 Jl '63.

(MIRA 16:10)

l. Belorusskiy nauchno-issledovatel skiy institut nevrologii, neyrokhirurgii i fizioterapii i Institut fiziki AN BSSR. Predstavleno akademikom AN BSSR D.A.Markovym.

1

PIKULIK, L.G.; GLAIXHENKO, L.F. Temperature effect on the luminescence of complex molecules in

various media. Izv. AN SSSR. Ser. fiz. 27 no.6:758-762 Je (MIRA 16:7) 63.

1. Institut fiziki AN Belorusako, SSR. (Molecular spectra)

8/0048/63/027/006/0758/0762

L 9860-63 EMP(j)/EPP(c)/EMT(1)/EMT(m)/EDS-AFFTC/ASD/ESD-3/SSD-Fr-4-MI/M/MI/IJP(C) ACCRESION NR: AP3001354

Pikulik, L. G.; Gladchenko, L.

TIME: Influence of temperature on the luminescence of complex molecules in different media [Report of the Eleventh Conference on Luminescence held in Minsk from 10 to 15 September 1962]

SCURCE: AN Seen. Izv. Seriya fizicheskaya, v. 27, no. 6, 1963, 758-762

TOPIC TAGS: luminescence of complex molecules, phthalimides, absorption of

ABSTRACT: Investigation of the temperature dependence of different characteristics of luminescence is important from the standpoints of elucidating the properties of fluorescing molecules and determining the nature of intermolecular interactions in condensed media. The solvents used in such investigations may be divided into two classes: those vitrifying and those crystallizing at low temperatures. In the present work there are considered the temperature dependences of a number of phthalimides in vitrifying solvents.

Card 1/3

L 9860-63 ACCESSION NR: AP3001354

Most of the phthalimides were identical with those investigated earlier by the authors; the solvents were ethyl and isobutyl alcohols, glycerol, n-hexane, n-octane, dioxane, ethyl ether and chlorobenzene. A major difficulty in obtaining absorption spectra stems from the fact that upon freezing the solution form a snow-like mass in which scattering is comparable to absorption; hence the absorption spectra were obtained by the method of diffuse reflection. For phthalimide and N-methylphthalimide some structure is discernible in the absorption spectra at room temperature; upon cooling, instead of becoming sharper, the structure is smeared out; no structure was observed for the other investigated phthalimides. On the basis of the temperature dependences the substances can be divided into two classes: 1) those in which both the absorption and fluorescence peaks exhibit a red shift upon cooling and 2) those in which the fluorescence changes, while the absorption spectrum remains the same. For most but not all the investigated substances the reciprocal of the polarization varies linearly with the temperature divided by the viscosity. The quantum efficiency and lifetime of the excited state appear to vary in the same way with temperature. "In conclusion the authors express their gratitude to V. V. Zelinskiy and I. I. Reznikova for making the phthalimides available and I. I. Reznikova for making the phthalimides available and to V. D. Tkachev and I. L. Belayts for performing some of the measurements of the temperature dependences of the excited Cord 2/3

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240

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| SUBITYED: 00 | | | -D1 ACRUEMY | | | |
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GURIMOVICH, G. F.[Hurynovich, H. P.]; PIKULIK, L. G.[Pikulik, L. H.];

SOLOV'YEV, K. M.[Salauiou, K. M.]

Anton Mikifaravich Seuchanka; on his 60th birthday. Vestsi AM
BSSR. Ser. fiz.-tekh. nav. no.1:124-128 '63.

(MIRA 16:4)

(Seuchanka, Anton Mikifaravich, 1903-)

GLADCHEMEO, L. F. [Hladchanka, L. F.]; PIKULIK, L. G. [Pikulik, L. H.]

Study of the electron spectra of organic molecules in scattering media. Vestsi AM BSSR. Ser. fiz.-tekh. nav. no.1: 52-58 '63.

(Molecular spectra) (Organic compounds)

PIKULIK, L.G.; SNOPKO, V.N.

Fluorescence polarization of complex molecules. Lok1. AN 568A 5 no.3:155-158 Mr '62. (EIRA 15:3)

1. Institut fiziki AN BSSR. Predstavleno akademikom AN BSSR A.N.Sevchenko.

(Fluorescence) (Polarization (Light)) (Molecular spectra)

AUTHORS

Dreytser, F F and Pikulik, L G.

\$ 250 62 006 009 002 004

1046/1246

TITLE:

Theoretical analysis of the temperature shift of electron spectra of complex molecules in

solutions

Akademiya nauk BSSR Doklady, v 6, no 9, 1962, 560-562 **PERIODICAL**

TEXT: McRae's theoretical conclusions concerning the effect of solvents on the electron spectra are com plemented with expressions allowing for the interaction at various temperatures between the medium and a fluorescing molecule. Contrary to the theoretical results, the dependence



$$v_{0-0}^{\bullet} - v_{0-0}^{f} = f \left(\frac{\tau}{\tau + \tau_{r}} \right). \tag{6}$$

where v_{0-0}^{s} , v_{0-0}^{f} — the electron transition frequencies in absorption and emission spectra, respectively τ — the average life-time of an excited molecule, t, relaxation time of a polar molecule, is not linear in the general case, probably due to the failure to take the temperature dependence of ε and n into account. There is 1 figure

ASSOCITION

Institut fiziki AN BSSR (Institute of Physics, AS BSSR)

PRESENTED:

By B. I. Stepanov, Academician, AS BSSR

SUBMITTED

February 17, 1962

Card 1/1

FIKULIK, L.G.; SOLOMAKHO, M.A.

Effect of temperature on the electronic spectra of complex molecules.
Opt. 1 spektr. 8 no.3:338-341 Mr '60. (MIRA 14:5)

(Spectrum, Molecular)

LJP(c)/SSD/BSD/SSD(c)/AFWI/ASD(a)-5/AS(mp)-2/ 20225-65 EWT(1)/EEC(b)-2 8/0250/64/008/010/0641/0614 APGC(b)/ESD(gs)/ESD(t)

AP5001200 ACCESSION MR:

AUTHOR: Pikulik, L. G.; Gladchenko, L. F.

STITLE: Determination of the constant dipole moments of excited molecules from the temperature shift of the fluorescence and absorption spectra

SOURCE: AM BSSR. Doklady, v. 8, no. 10, 1964, 641-644

TOPIC TAGS; dipole moment, excited molecule, fluorescence spectrum, obsorption spectrum, line shift

ABSTRACT: The formula used to calculate the dipole moment is

$$v_{00}^{a} - v_{00}^{b} = \Delta v_{00}^{a-1} = \frac{2(\vec{p}_{0} - \vec{p}_{1})^{2}}{hca^{2}} \left[\frac{s-1}{s+2} - \frac{n^{2}-1}{n^{2}+2} \right] \frac{\tau}{\tau + \tau_{\tau}},$$

and is derived on the basis of an earlier theoretical description of the temperature shift of the electronic spectra of complex molecules in dipole solvents (E. F. Dreytser and L. G. Pikulik, DAN BSSR v. 6, 560, 1962). In this formula 700 -- Frequencies of the electronic transitions for absorption (a) and fluorescence (f), μ_0 and μ_1 -- constant dipole moments of the unexcited and excited

Card 1/3

L 20225-65 ACCESSION NR: AP5001200 2

molecule, respectively, a -- static dielectric constant of the medium, n -- refractive index of the medium extrapolated to zero frequency, a -- parameter of the given molecule, c -- velocity of light, h -- Planck's constant, I -- lifetime of the excited molecule, and Ir -- relaxation time of the solvent molecule. An analysis of this formula shows that the frequency difference becomes appreciable on going from low temperatures to room temperatures, owing to orientational effects. The calculations of the dipole moments by this formula were made by using fects. The calculations of the dipole moments by this formula were made by using data on the temperature shift of the spectra in isobutyl alcohol in the temperature range from + 20 to - 196C. The results are listed in Table I of the enclosure. The data are compared with those obtained by others and some of the causes for some of the discrepancies are discussed. The authors thank Academician AN BSSR A. N. Sevchenko for continuous interest in the work. This report was presented by A. N. Sevchenko. Orig. art. has: I table and I formula.

ASSOCIATION: Institut fiziki AN BSSR (Institute of Physics AN BSSR)

SUBMITTED: 29Dec63

ENCL: Ol

SUB CODE: OP

MR REF SOV: . 008

OTHER: 003

Card 2/3

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| Substance | A*00 4m | 77,-77 L | 3,7 | 7,04-9 | <u> </u> |
| 4-acetyl-anino-M-methyl pithalinide 4-againo-M-methyl-phthalimide 4-againo-M-methyl-phthalimide | 2500 3800 4100 | 5,2 6,3 8,56 | 4.6 3.6 3,7 | 9,8 9,9 10,26 | 9,5 9,7 10,0 |
| archylphthalanide archylphthalanide 4. disethyl-anino-M- acthylphthalanide 3.esetyl-anino-M- acthylphthalanide 3.esetyl-anino-M- acthylphthalanide 3.eseino-M-methylphthalanide | 5400 700 | 7,45 2,7 4,8 | 3,7 , 2,4 2,7 | 6.1 7.2 | 3,80 |
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APANASEVICH, P.A.; BORISEVICH, N.A. VOLOD'RO, L.V.; GLALCHENRO, L.F.;
GRIBKCVSKIY, V.F.; GUALLOVICH, G.F.; IVANOV, A.F.; RVZAMIDOVA,
V.V.; FIKULIK, L.G.; FILLIFOVICH, L.A.; RUBANOV, A.S.; FYBANOV,
V.S.; SAMSON, A.M.; SARZHEVSKIY, A.M.; SCLOV'YEV, K.N.;
UMREYKO, D.S.; KHAFALYUK, A.F.; YELLVASHEVICH, M.A., akademik,
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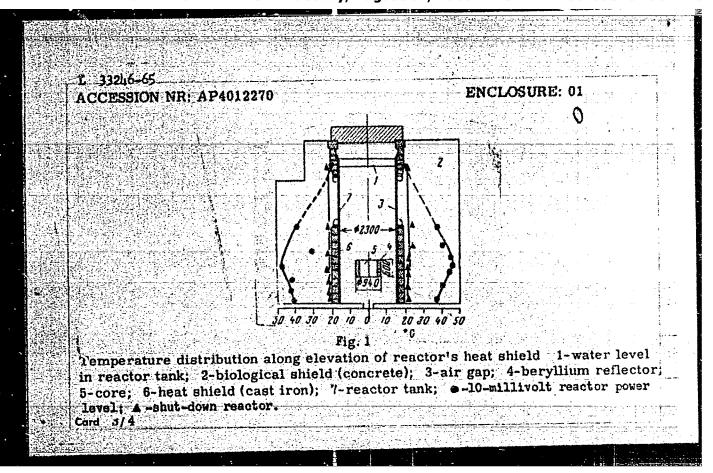
[Interaction between nonequilibrium ratiation and matter Vzaimodeictvie neravnovesmore politichemia sive, hehestvom. Minsk, Nauka i tekmika, 1965. 223; (11x4 18:3)

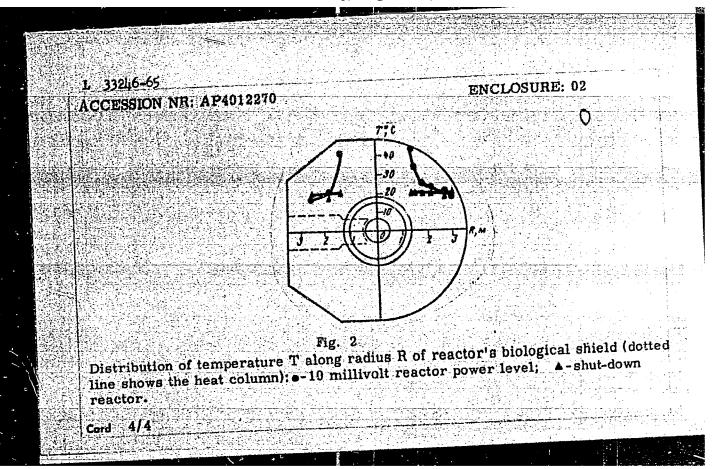
1. Akademiya nauk SSSR. Institut fiziki. Akademiya nauk Belorusskoy SSR (for Yellyashevich).

ENT(m)/ETF(c)/EPF(n)-2/ENG(m)/EPR Pr-4/Ps-4/Pu-4 DM 1 33246-65 S/0089/64/016/001/0068/0069 ACCESSION NR: AP4012270 AUTHOR: Konopley, K. A.; Pikulik, R. G. TITLE: Measurement of the temperature distribution in the reactor VVR-M shield SOURCE: Atomnaya energiya, v. 16, no. 1, 1964, 68-69 TOPIC TAGS: reactor shield, temperature distribution, thermal column graphite, nuclear power reactor/ VVR-M reactor ABSTRACT: The authors describe the construction of the heat shield and of the concrete shield of the VVR-M reactor and of the cooling system. The thermocouples used for temperature measurements were placed in the shield in different locations before the concrete was set. The temperature of the graphite of the thermal column was measured by a thermocouple inserted during the experiment. The temperature distribution curves have a maximum at the level of the axis of the reactor cone. On the horizontal axis the temperature drops sharply at the transition from iron to concrete. The hottest point of the iron shield is 49C. Card 1/4

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 CIA-RDP86-00513R001240

| ACCESSION NR: AP4012270 | | | | | | | |
|---|----|--------------|------------|---|--|--|--|
| When the reactor power is at 10MW, the maximum temperature of the graphite of the thermal column is about 200C. "The authors are grateful to the scientific supervisor <u>D. M. Kaminker</u> for his help." Orig. art. has: 3 figures | | | | | | | |
| ASSOCIATION: None | | | | • | | | |
| SUBMITTED: 05Mar63 | | | ENCL: 02 | | | | |
| SUB CODE: NP | NR | REF SOV: 000 | OTHER: 000 | | | | |
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L 26243-66 EWA(h)/EWT(1)

ACC NR. AP6013503

SOURCE CODE: UR/0120/66/000/002/0088/0089

AUTHOR: Pikulik, V. G.; Pikulik, R. G.

43 B

ORG: Belorussian State University, Minsk (Belorusskiy gosudarstvennyy universitet)

TITLE: Transistorized generator of nanosecond pulses

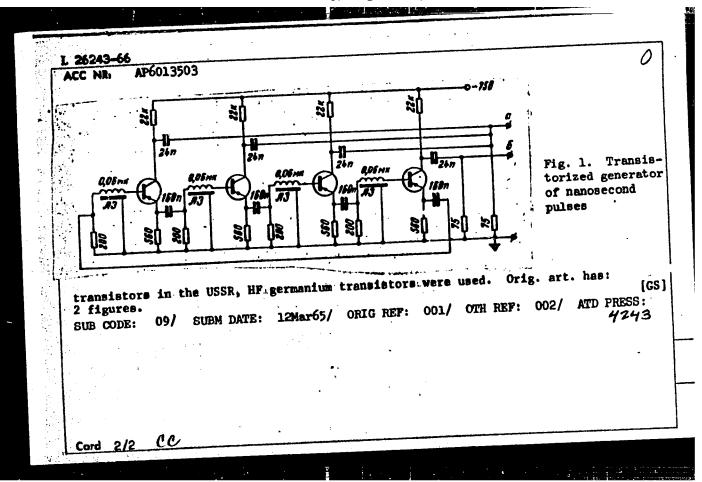
SOURCE: Pribory i tekhnika eksperimenta, no. 2, 1966, 88-89

TOPIC TAGS: pulse generator, nanosecond pulse, pulse rate, HF transistor

ABSTRACT: The special feature of the described generator is that each transistor has its own dynamic bias and a common delay line. Sections of this line with the transistors form a ring system which generates pulses with identical periods. A circuit of the four-stage ring generator is shown in Fig. 1. The circuit incorporates four the four-stage ring generator operating at avalanche condition and utilizes sections of Ph06 germanium transistors operating at avalanche condition and utilizes sections of the RKZ-400 cable (z = 400 ohms) as a delay line. The circuit generates pulses with an amplitude of 1.5 v, a duration of 30 nanosec, and a prf of 13.5 Mc. The pulse generator may be loaded by individual loads to provide time-space distribution of output pulses. It is stated in conclusion that owing to the lack of avalanche

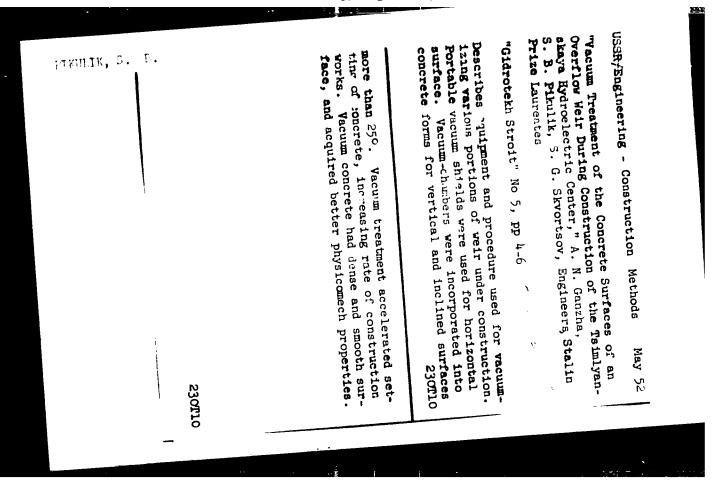
Cord 1/2

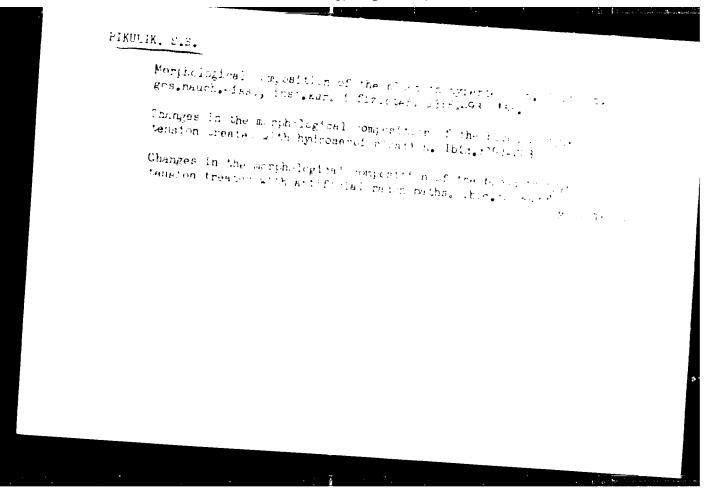
UDC: 621.373.5



PIKULIN, S.A.; SHCHERBAKOV, V.A.; DONSKAYA, S.N., GORCKHOV, L.S. Y-ray diffraction study of the phase composition of open hearth slags during the smelting period. Zev. lab. 30 no.9:1102-(MIRA 18. 1) 1105 '64. 1. Moskovskiy institut stali i splavov.

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SWEKHTER, S.Ya.; REMNITSKIY, A.M.; PIKULIN, S.A.; KREMWAK, E.M.
       automatic walning of oversize, steel pouring ladles. Avtom.s.ar.
                                                               (MIRA 18:3)
       18 :0.1:59-60 Ja *65.
       . Mommunarskiy metallurgicheskiy zavod (for Shekhter, Revniteki,
       Powelln). The titut elektrosvarki im. Ye.O. Patona AN Ukraik (for
       fir menko).
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FIKULIK, J.J.

36877. C monotsitarnov reaktsii u bol'nykh gipertonicheskoy bolezn'yu. Lechennykh fizicheskimi metodami. Trudy Uzbak. gos. Nauch.-issled. in-ta kurortoloii i fizicterajii im. Jemashko, sb.11, 1949, c. 232-45

SO: Letopis' Zhurna Nykh Statey, Vol. 50, doskym, 1949

AUTHORS:

Starobinets, G. L., Pikulik, V. A.

76-32-4-3/43

TITLE:

The Structure of the Adsorption Layer at the Interface Between Air and Concentrated Solution. II. (Stroyeniye adsorbtsionnogo sloya na granitse kontsentrirovannyy nevodnyy rastvor -

PERIODICAL:

Zhurnal Fizicheskoy Khimii, 1958, Vol. 32, Nr 4, PP. 756-759 (USSR)

ABSTRACT:

In a previous paper it was shown that the possibility of the formation of a polymolecular adsorption layer in greatest in two cases: When the energy of the intermolecular interactions 1-2 is relatively small compared to those between 1-1 or 2-2. When the molecules of the active surface component form groups with parallel oriented dipoles. In order to investigate the first case the authors investigated in this paper the system benzene-polar derivatives of benzene (benzyl alcohol phenol, nitrobenzene and chlorobenzene) with the active benzene acting as surface component. The thermodynamic benzene activity as well as the dielectric permeability and density

(at 70°C) were determined for this system by Martin (ref. 2). A table of the additional measurements of the surface tensions

Card 1/3

The Structure of the Adsorption Layer at the Interface Between Air and Concentrated Solution. II.

76-32-4-3/43

at 70°C as well as of the calculated adsorption in the air--solution interface, and of the dielectric polarization according to Onsager-Kirkwood and of the correlation parameter of the polar component according to Kirkwood is given. From the graphically given results it can be seen that the maximum adsorption of benzene in benzyl-alcohol is abnormally great and decreases according to the series of solvents phenol-nitrobenzene-chlorobenzene. The diagrams of the thermo dynamic benzene activity composition show that the deviation from the ideal case as well as the maximum quantity of benzene adsorption take place sympathetically. The abnormally great benzene adsorption is explained by the group formation of molecules of the surface-inactive components. Investigations of the correlation parameter of the polar components showed that the concentration of dimers in the system benzylalcohol--benzene is greater than in phenol-benzene, in the former the depth of the zone of negative adsorption of the polar components being greater. The distribution coefficients of benzene between the surface and the center of the solvent were calculated according to the theory by A. A. Zhukovitskiy, and were represented graphically. It is shown that the

Card 2/3

"APPROVED FOR RELEASE: Tuesday, August 01, 2000 The Structure of the Adsorption Layer at the Interface Between Air and Concentrated Solution. II. adsorption layer of benzene in nitrobenzene and chlorobenzene is monomolecular, and in phenol and benzylalcohol 76-32-4-3/43 polymolecular. Finally the authors state that the abnormal properties of the adsorption layers found are to be traced pack to a bolymolecniar character as well as a conditioned brobarrias or rus snaorbrion rahars round are no be character adsorption depth which is determined by the mean length of There are 4 figures, 2 tables, and 5 references, 3 of which are Soviet. ASSOCIATION: Belorusskiy gosudarstvennyy universitet im. V. I. Lenina Minsk (Minsk, Belorussian State University imeni V. I. SUBMITTED: May 21, 1956 AVAILABLE: Library of Congress Card 3/3 1 Adsorption--Theory properties 3 Benzene--Adsorptive properties 2 Benzene derivatives--Adsorptive

100-100

Puvipuoneimo,, Duocyta, r... AUTHORSE

Spectrochemical determination of the spectrochem 111111

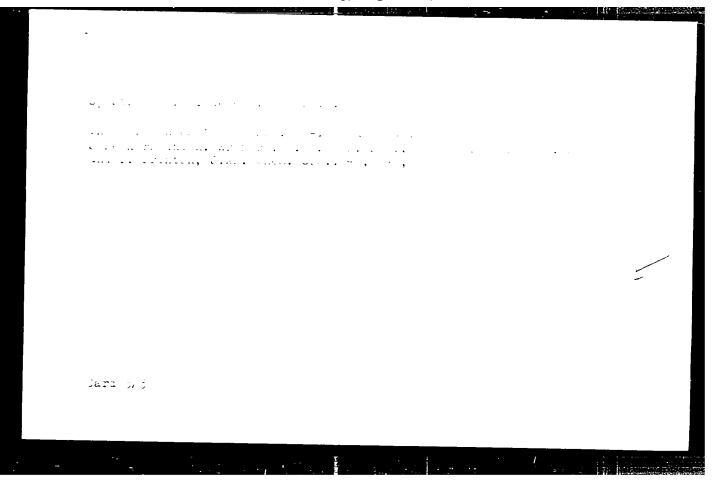
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Mansk, 1700, 45 - 10

The present work was alled at letterality of earths in a number of agatites and phosymorites against others of methods, as these are . . . Level hore promising than the curvent. ones. For qualitative entimation, the sample war of the el carbon (1:1 ratio) end the diestra were justic minde de (IUP-91) spectro, raja, usia, tak - ---- ...- - curro . . . 4500 a region. Experimentally actual are process and in the were found in the apatites and in one insagnorite, while ing phosphorites required chemical concentration before the copic analysis. All the lanthunons were actected. For the amitting

Card 1/ 3



38599-65 EWT(1)/EEC(k)-2/EWG(m)/T/EEC(b)-2/EWA(h) Pm-4/Pz-6/Peb IJP(c)

ACCESSION NR: AP5005984

5/0108/65/020/002/0052/0056

AUTHOR: Pikulik, V. G. (Active member); Shats, S. Ya. (Active member)

TITLE: Avalanche properties of industrial alloy-junction low-power transistors

SOURCE: Radiotekhnika, v. 20, no. 2, 1965, 52-56

TOPIC TAGE transistor, alloy junction transistor, low power transistor, industrial transistor

ABSTRACT: The psculiarities of behavior of alloy transistors at high collector voltages are analyzed. The breakdown voltage and some other characteristics of P12-P407, P27-P28, P13-P16, P25-P26 transistors are tabulated. R-f types P12, P406, and P407 are recommended as most suitable for operating under avalanche conditions. Transistors with a collector-junction breakdown voltage close to the estimated average breakdown voltage U* should be used for avalanche application. Those transistors have pronounced avalanche characteristics whose

Card 1/2

| transistors. P101-P103 sili | n develops in the center ter-base may be used for con transistors have ve | or avalanche-wise selecting of ry nonuniform junctions. | |
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| Often they exhibit an interlin some Pl6B and Pl1 transist 1 table. | | | |
| | and the contract of the contra | arter 🖎 lart i Genet difficio de la facilità del Education della discussione della consecuencia della conse | 1 |
| | | o radiotekhniki i elektrosvyazi ng and Electrocommunication) SUB CODE: EC | |
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L 26243-66 EWA(h)/EWT(1)

ACC NR. AP6013503

SOURCE CODE: UR/0120/66/000/002/0088/0089

AUTHOR: Pikulik, V. G.; Pikulik, R. G.

ORG: Belorussian State University, Minsk (Belorusskiy gosudarstvennyy universitet)

TITLE: Transistorized generator of nanosecond pulses

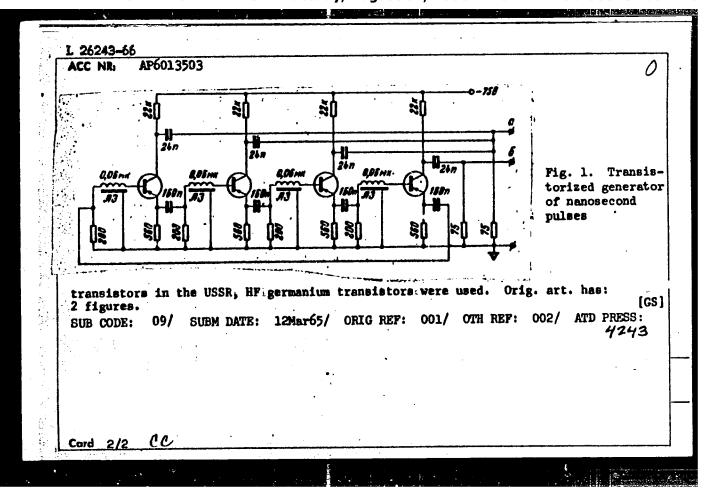
SOURCE: Pribory i tekhnika eksperimenta, no. 2, 1966, 88-89

TOPIC TAGS: pulse generator, nanosecond pulse, pulse rate, HF transistor

ABSTRACT: The special feature of the described generator is that each transistor has its own dynamic bias and a common delay line. Sections of this line with the transistors form a ring system which generates pulses with identical periods. A circuit of the four-stage ring generator is shown in Fig. 1. The circuit incorporates four P406 germanium transistors operating at avalanche condition and utilizes sections of the RKZ-400 cable (z = 400 ohms) as a delay line. The circuit generates pulses with an amplitude of 1.5 v, a duration of 30 nanosec, and a prf of 13.5 Mc. The pulse generator may be loaded by individual loads to provide time-space distribution of output pulses. It is stated in conclusion that owing to the lack of avalanche

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LDC: 621.373.5



PIKULIK, V.G.; SHATS, S.Ya.

Avalanche characteristics of industria. Non-power of the transistors. Padioteknnika 20 no.2. \$2.50 in the contract of the cont

1.Deystvitelingye chleny Nauchno-tekiriceski komponistre radiotekhniki i elektrosvyazi imeni romaka

ACCESSION NR: AP4037399 S/0106/64/000/005/0046/0052

AUTHOR: Pikulik, V. G.; Shats, S. Ya.

TITLE: Using the avalanche characteristics of junction transistors in pulsed

devices

SOURCE: Elektrosvyaz¹, no. 5, 1964, 46-52

TOPIC TAGS: transistor, avalanche transistor, P406 transistor, P407 transistor, avalanche transistor pulse generator

ABSTRACT: A review of practical circuits operating with avalanche transistors (Soviet types P406 and P407) is presented. Capacitor-type 1-mc and 5-mc relaxation oscillators with pulse-repetition frequency stabilized by delay lines were tested, as well as a 3-mc slave multivibrator and a 10-microsec pulse-packet generator with a fill frequency of 1.25 mc. A number of Soviet-make transistors were tested for 5,000 hrs in a simple avalanche-relaxation-oscillator

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ACCESSION NR: AP40 37 399

circuit; no appreciable change in parameters was detected. The avalanchetransistor circuits were taken from American sources (W. Shockley, et al., Proc. IRE, 1959, v. 47, no. 6, and elsewhere). Orig. art. has: 8 figures and 8 formulas.

ASSOCIATION: none

SUBMITTED: 18Jul63

DATE ACQ: 09Jun64

ENCL: 00

SUB CODE: EC

NO REF SOV: 001

OTHER: 003

Cord 2/2

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PIKULIK, V.G.; SHATS, S.Ya.

Relaxation oscillator on an avalanche transistor with grounded emitter and base. Radiotekhnika 18 no.11:57-61 N '63.

(MIRA 16:12)

1. Deystvitel'nyye chleny Nauchno-tekhnicheskogo obshchestva radiotekhniki i elektrosvyazi imeni Popova.

... III. ELECTRICAL

S/194/61/000/010/034/082 D256/D301

9.6000

Pikulik, V.G.

TITLE

.LTHOR:

Transistorized multi-position multi-channel contactless commutator for switching spectra of acoustic

frequencies

PERIODIC.A.:

Referativnyy zhurnal. Avtomatika i radioelektronika no. 10, 1961, 4, abstract 10 V34 (Nauchno-tekhn inform. byul. Leningr. politekhn. in-t. 1960, no. 9

45-52)

TEXT. The problems are considered of constructing a cyclic contactless commutator (CC) for switching spectra of acoustic frequencies used in pulse communication systems and computer techniques. CC consists of a set of key elements linked to the central control unit through individual control units of the key elements. With the key groups operating in sequence each CC output is successively connected to all the inputs of the CC. The basic requirement set upon

Card 1/2

\$/194/61/000/010/034/062 D256/D301

Transistorized multi-position...

the CG is the correspondence of the cl. characteristic of the CG (cross, linear and nonlinear distortions) to the characteristics of mechanical commutators. The sources of possible distortions are considered and principle diagrams of separate elements are given together with circuit diagrams of the control pulse generator and the control unit with one key element - design of operating a voltage of 3 V amplitude, the time of the full cycle being 0.5 sec age of 3 V amplitude, the time of the full cycle being 0.5 sec are no cross-distortions. Transistors of the N-11 (P-11) and N-14 (P-14)-type and N-2 (D-2)-type diodes are employed. The variations of the CG output levels during operation do not exceed 10-30 mV. The working temp, ranges from -40 to +50 G. 11, figures 3 references. Abstracter's note: Complete translation.

Card 2/2

PIKULIN, A.P., dotsent

Preliminary estimate of position errors for distant points of an underground theodolite traverse. Izv. vys. ucheb. sav.; gor. zhur. no. 11:111-116 '60. (MIRA 13:12)

1. Khar'kovskiy gornyy institut, Rekomendovana kafedroy marksheyderskogo dela Khar'kovskogo gornogo instituta. (Mine surveying)

PIKULIN, A. P.

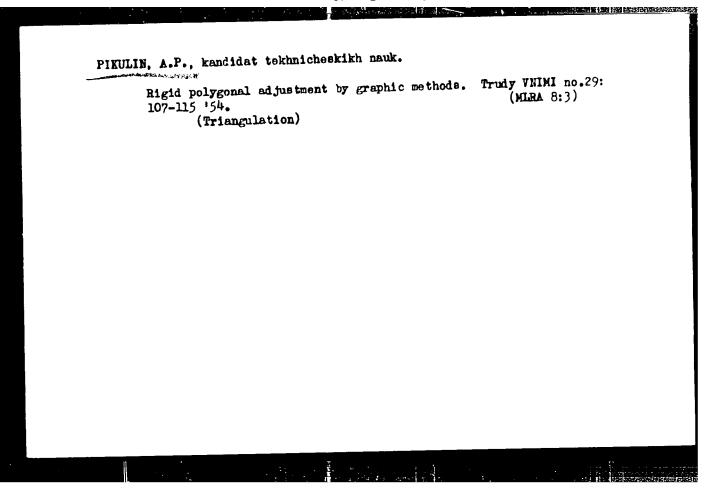
RYZHOV, Petr Aleksandrovich, prof., doktor tekhn.nauk; BUKRINSKIY, Viktor
Aleksandrovich, dotsent, kand.tekhn.nauk; GUDKOV, Valentin
Mikhaylovich, kand.tekhn.nauk; KROTOV, Gavriil Alekseyevich,
dotsent, kand.tekhn.nauk; LYUBMAN, Izrail' Borisovich, assistent;
HUDAKOV, Mikhail Lazarevich, prof., doktor; PIKULIN, A.P., kand.
tekhn.nauk, retsenzent; BUTKEVICH, T.V., red.; PARTSEVSKIY, V.N.,
red.izd-va; BEKKER, O.G., tekhn.red.

[Mine surveying] Marksheiderskoe delo. Pod nauchnoi red. P.A.
Ryzhova. Moskva, Gos. nauchno-tekhn.izd-vo lit-ry po chernoi i
tsvetnoi metallurgii, 1958. 463 p.
(Mine surveying)

LOGVINENKO, N.V., doktor geolgo-mineralogicheskikh mauk; PIKULIN, A.P., otvetstvennyy redaktor; ZABROISKIY, Kh.A., tekhnicheskiy redaktor.

[Lithology and paleogeography of the productive layer of the Donets Basin Carboniferous] Litologiia i paleogeografiia produktivnoi tolahchi donetskogo karbona. Khar'kov, Isd-vo Khar'kovskogo gos. univ. im. A.N.Gor'kogo, 1953. 434 p. [Microfilm] (MERA 8:2)

(Donets Basin--Geology, Stratigraphic) (Donets basin--Paleogeography)



PIKULIN, Saveliy Moiseyevich; INDENBAUM, V.S., redaktor; IANOVSKAYA, M.S., redaktor izdatel'stva; KARASEV, A.I., tekhnicheskiy redaktor

Control of the Contro

[New method of laying foundations for turbine installations] Novyi metod ustanovki na fundamente turboagregatov. Moskva, Gos. nauchnotekhn. izd-vo lit-ry po chernoi i tsvetnoi metallurgii, 1956. 39 p. (Turbines) (Foundations) (MIRA 9:12)

GURINOVICH, G.P.; PIKULIK, L.G.; SOLOV'YEV, L.N.

Sixth conference on luminiscence. Inzh.-fiz. zhur. no. 6:115-117

Je '58. (Luminiscence)

CZECHOSLOVAKIA

SPRINGER, V., PIKULIKOVA, 2.

Department of Analytical Chemistry, Faculty of Pharmacy, Comenius University (Katedra analytickej chemie Parmaceutickej fakulty University Komenskehe), Bratislava

Bratislava, Farmacouticky obser, No 10, October 1965, pp 418-427

"Method of incineration of organic substances in a flask and its use in pharmaceutical control. Part 4: Determination of chlortet-racycline and chloremphanical."

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RUTKOVSKIY, F., pensioner; KURHOSOV, M.; DEOZDOV, V.; PIKULIN, F.(Gor'kiy);

We offer the following solution. Sov.profsoiusy 7 no.23:
(MIRA 12:12)

1. Tekhnicheskiy inspektor Mosoblprofsoveta (for Kurnosov).
2. Instruktor Belorusskogo respublikanskogo soveta profsoyuzov (for Drozdov).
(Labor laws and legislation)
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THE REMINERATIONS

Leaders in communist competition. Mashinostroitel' no.4: 40-41 Ap '60. (MIRA 13:6) 1. Predsedatel' profkoma kuzovnogo korpusa Gor'kovskogo avtozavoda. (Gorkiy--Automobile industry-Technological innovations) (Socialist competition)